Polymer Bulletin 5, 255-261 (1981)

Polymer Bulletin © Springer-Verlag 1981

¹³C NMR Spin-Lattice Relaxation in Solid Poly(Methyl methacrylate) **Effects of Tacticity and of Plasticizers**

Hommo T. Edzes and Wiebren S. Veeman

Department of Molecular Spectroscopy, Faculty of Science, University of Nijmegen, Toernooiveld, 6525 ED Nijmegen, The Netherlands

Presented at the 22nd Microsymposium, "Characterization of Structure and Dynamics of Macromolecular Systems by NMR Methods", Prague, July 20-23, 1981

SUMMARY

 ^{13}C spin-lattice relaxation rates R_1 in samples of solid PMMA have been measured with CP MAS NMR. Isotactic, atactic, syndiotactic, and the 1:2 i-s PMMA complex were measured, along with atactic samples containing bisdioxan or the monomer as a plasticizer. The reorientation of the α methyl protons is the dominant relaxation source for the methylene, α methyl and quarternary carbon nuclei. Increased mobility of the polymer in the presence of a plasticizer is not reflected in significant changes in R₁.

INTRODUCTION

The combination of the NMR techniques of $1^{3}C-\{1_{H}\}$ cross-polarization (CP) (PINES et al., 1973), of fast magic angle spinning (MAS), and of strong dipolar decoupling (DD) of the protons during acquisition of the ¹³C FID signal, has made it possible to obtain high-resolution ¹³C NMR spectra of polymers in the solid state (SCHAEFER and STEJSKAL, 1976; LYERLA, 1979; GARROWAY et al., 1979). The spectra are usually well resolved, and the spin-lattice relaxation rate can be measured for each individual resonance peak. In particular, both the relaxation of the carbon atoms on the polymer backbone and of those on side chain groups can be followed separately. Since the spin-lattice relaxation rates are sensitive to local motions, it is possible, in principle, to probe separately the motions of the side chain groups and of the polymer backbone.

The mechanical properties of a solid polymer are related to motions of the polymer molecules. Both the tacticity and the presence of a plasticizer have a profound effect on the mechanical properties of poly(methyl methacrylate), PMMA. We here report the effect of tacticity and of plasticizers on the $^{13}\mathrm{C}$ spin-lattice relaxation rates R₁ in solid PMMA. R₁=T₁⁻¹, where T_1 is the spin-lattice relaxation time. The main effect of a plasticizer is expected to be an enhanced mobility of the polymer chain. R1 values are most sensitive to motions which occur at a rate comparable to the NMR frequency, i.e. to motions in the range of 107-10¹⁰ Hz. Therefore, the measurement of spin-lattice relaxation rates in the rotating frame, R10, might seem to offer a more sensitive way to probe changes in the mobility of the polymer backbone. However, if $R_{1\rho}$ is measured under CP MAS NMR conditions, no simple and straightforward relation exists with the slower molecular motions (GARROWAY et al., 1979; VanderHART and GARROWAY, 1979). For that reason, R_{10} measurements are not included here.



FIGURE 1. ¹³C CP MAS NMR spectra of solid PMMA at room temperature. The peaks between 90 and 130 ppm are spinning sidebands; black peaks in the α -CH₃ region are ascribed to trapped solvent molecules. The stick spectrum shows the resonance positions in solution. i: isotactic dyad (m) or triad (mm) resonances; s: syndiotactic dyad (r) or triad (rr) resonances; broken lines: heterotactic triad (mr) resonance.

EXPERIMENTAL

 13 C NMR spectra and spin-lattice relaxation rates were measured at ambient temperature with a home-built CP MAS NMR spectrometer, operating at a 13 C NMR frequency of 45.27 MHz. The pulse sequence programmer (Van OS and VEEMAN, 1979) and the design of the sample spinner accessory (Van DIJK et al., 1980) have been published elsewhere; the doubly-tuned single-coil probe design of STOLL et al. (1977) is used. Cross-polarization and proton dipolar decoupling were achieved with 13 C and 1 H RF magnetic field strengths of 35 kHz. A CP contact time of 1 ms was used, and the sample spinning rates ranged from 2.4 to 3.0 kHz.

 13 C spin-lattice relaxation rates were measured with a slightly modified version of the TICP method described by TORCHIA (1978). This method has distinct advantages over the inversion-recovery method for the measurement of slowly relaxing 13 C resonance: the final equilibrium magnetization does not need to be measured, since the TICP relaxation decay tends to zero for long relaxation delays; and, a slowly relaxing background signal from the Kel-F spinner and the coil surroundings, which hampers an accurate determination of peak amplitudes in the inversion-recovery experiment, is completely absent in the TICP relaxation delay time by applying a train of short pulses at the proton NMR frequency (2 μ s repetition rate; 15% duty cycle).

Spin-lattice relaxation rates were determined from a least-squares fit of the measured peak amplitudes to the relaxation function $A(t)=A(0) \exp(-R_1t)$. Some of the relaxation curves consistently showed non-exponential behavior; in those cases we choose to characterize the decay with the function $A(t)=A(0) \exp(-R_1t(1+fR_1t))$ (BAKAY et al., 1975). This expression is useful to describe slightly non-exponential curves, as long as very long relaxation delay times t are not included. With this function a good fit was obtained to all the measured points of the non-exponential relaxation curves. The spin-lattice relaxation rate R_1 from such a fit describes the <u>initial</u> slope of the relaxation decay; the non-exponentiality factor f characterizes the degree of non-exponentiality.

The PMMA samples studied were of high molecular weight and are listed in TABLE I. The tacticity was determined from ¹H spectra of 5% w/v solutions in CDCl₃. Bisdioxan and methyl methacrylate were incorporated as a plasticizer. Commercial Perspex was studied in the form of a massive spinner. All other samples were studied in powdered form and packed into Kel-F spinners with a sample volume of .28 $\rm cm^3$. About 2000 scans were needed to obtain a good signal-to-noise ratio.

RESULTS AND DISCUSSION

¹³C NMR spectra. Typical ¹³C CP MAS NMR spectra of the PMMA samples are shown in FIGURE 1. The plasticizer molecules (bisdioxan or methyl methacrylate monomer) did hardly crosspolarize under our experimental conditions; we ascribe this to the large mobility of the plasticizer molecules.

The CP MAS ¹³C NMR resonances of PMMA are broad compared to those measured in solution. For comparison, a stick spectrum of the resonance positions of PMMA in solution is included in FIGURE 1 (PEAT and REYNOLDS, 1972; LYERLA et al., 1977). The four functional groups of PMMA are easily assigned, although the CCH₂ and -OCH₃ resonances overlap. The resonance positions agree with those measured in solution. Fine structure due to

Sample	Tacticity	Source		Re	laxation	rates R_{l} (s ⁻¹)	
	(mm/mr/rr triad ratio)		α-cH ₃).C=0	5cH ₂	-0CH3	202
Commercial Perspex	4/31/65		19	.10	.10	.16 (f=.07)	.25 (f=.07)
Atactic PMMA	similar	-	20	.10	.11	.16 (f=.05)	.25 (f=.07)
* Ibid., + 20% w/w bisdioxan			1	.11	.12	.18 (f=.07)	.28 (f=.07)
* Ibid., + 40% w/w bisdioxan		-	21	.12	.16	.18 (f=.03)	.29 (f=.06)
* Ibid., + 10% w/w monomer		2	ł	.11	.14	.19 (f=.06)	.28 (f=.08)
Isotactic PMMA	95/5/0	Э	10	.06	60.	.14 (f=.11)	.17 (f=.11)
Syndiotactic PMMA	1/9/89	З	19	.10	.10	.17 (f=.07)	.28 (f=.09)
1:2 i-s PMMA complex	as above	¢	11 (i) 19 (s)	.09 (i) .13 (s)	.12	.19 (f=.09)	.27 (f=.08)
Estimated error in the relax:	ation rates		107	15%	30%	10% (f:50%)	10% (f:50%)
<pre>f characterizes a non-exponen * Bisdioxan = 2,4,8,10-tetrao</pre>	ntial relaxation (xaspiro-[5.5]-unde	see experii cane:	mental sec χ_0^0	tion)			

samples.
PMMA
solid
some
of
MHz
45.27
at
rates
relaxation
attice
spin-1
13 _C
TABLE

PMMA Sources: 1. Nijverheidsorganisatie TNO, Delft, The Netherlands

Monomer = methyl methacrylate, MMA

2. Philips Research Laboratories, Eindhoven, The Netherlands

3. Department of Polymer Chemistry, University of Groningen, The Netherlands

differences in the tacticity is not resolved very well: only the isotactic (mm) and syndiotactic (rr) triad resonances of the >C=0 and α -CH₃ group are separated well enough to become identifiable. This is very clear in the spectrum of the stereocomplex of iso- and syndiotactic PMMA. Non-resolved fine structure is a major source of the line broadening of the resonances. In particular the α -CH₃ resonance is rather broad and clearly asymmetric, even in the spectra of the stereoregular samples. One cannot simply assign the three triad resonances; instead, a careful inspection of the line shape seems to hint at a larger number of overlapping lines. We are currently investigating the α -CH₃ region in detail, hoping to identify specific conformations in the solid state. Both the folding of the main chain, and the orientation of the ester group, can give rise to a multitude of lines in the α -CH₃ region.

Spin-lattice relaxation. The measured relaxation rates R_1 are tabulated in TABLE I. The relaxation of the -OCH₃ and CC carbons proceeds distinctly non-exponential over the experimental range of relaxation delay times t (16 s. maximum). For these lines the non-exponentiality factor f is given as well (see experimental section). α -CH₃ rates were measured from the largest peak amplitude. From changes in the line shape it is clear, however, that the α -CH₃ relaxation can be decomposed into three differently relaxing components of equal width, and centered at 16, 20, and 23.5 ppm. Relaxation rates of 20, 13, and 19 s⁻¹ are measured, with relative intensities of 4:2:1 respectively. It is tempting to identify these components with the α -CH₃ reisonances. However, the line shapes differ in detail from a simple three-line spectrum, again indicating a multitude of overlapping α -CH₃ resonances. Moreover, the intensity of the component at 23.5 ppm is much larger than is expected for the isotactic (mm) resonance in atactic PMMA.

From TABLE I it is clear, that tacticity has the largest effect on the spin-lattice relaxation rates R_1 . The relaxation proceeds considerably slower in isotactic than in syndiotactic PMMA, both in the pure stereoregular samples and in the stereocomplex. Commercial perspex and the atactic PMMA samples behave similar to syndiotactic PMMA. This is to be expected, since samples prepared by free radical polymerization are predominantly syndiotactic. The presence of plasticizers hardly affects the relaxation rates. At most, a slight trend towards faster relaxation occurs.

¹³C spin-lattice relaxation rates of solid PMMA have been published before. Prelimenary data given by VEEMAN and MENGER (1981) compare to those given here; the present data are more accurate. Spin-lattice relaxation in nonspinning PMMA has been measured by SCHAEFER et al. (1975). Their results differ from the ones presented here.

Comparing the 13 C relaxation data with those obtained in solution (LYERLA et al., 1977; HATADA et al., 1977), the most significant change occurs in the CH₂ relaxation: in solution, R₁ is about two orders of magnitude larger than in the solid. This clearly reflects the much larger mobility of the polymer backbone in solution. In the solid state, the presence of a plasticizer seems to have the largest effect on the CH₂ relaxation. This might arise from an increase in the backbone mobility. Unfortunately the accuracy is not good enough to be very definite about the effect of a plasticizer.

The α -CH₃ relaxation proceeds rapidly. R₁ is close to the theoretical maximum of about 30 s⁻¹ for a methyl group (DODDRELL et al., 1972), indicating that the α -CH₃ group reorients at a rate comparable to the NMR

(angular) frequency of 3. 10^{+8} s⁻¹. Both in solution and in the solid state, the α -CH₃ relaxation proceeds faster in the syndiotactic than in the isotactic conformation. From other experiments it is known, that the barrier to rotation for the α -CH₃ group is lower in the isotactic conformation, resulting in faster rotation (see LYERLA et al., 1977). Faster rotation, leading to slower relaxation, implies that the α -CH₃ rotation is to the high-temperature side of the R₁ maximum. The α -CH₃ relaxation data are compatible with proton spin-lattice relaxation measurements, where it was shown that the R₁ maximum at a similar NMR frequency occurs near 0°C (SLICHTER, 1966).

In the solid state, the rotation of the α -CH₃ group seems to completely dominate the relaxation of the other carbon nuclei, except for the -OCH₃ carbon. This can be concluded from the observed ratio of the relaxation rates. Let us assume that the reorienting methyl protons constitute the only source for dipolar relaxation. The carbon relaxation rates then should be proportional to r_{CH}^{-6} , where r_{CH} is the carbon-proton distance. The expected R_1 of the CC carbon, relative to the one of the α -CH₃, follows from a straightforward calculation. r_{CH} equals 2.15 A and 1.09 A, respectively. Thus, the CC carbon should relax at a rate smaller by a factor of 60, which is in excellent agreement with the measurements. For the CH_2 and the C=0 carbons the calculation of r_{CH}^{-6} is not as accurate, because the result depends on the assumed conformation. Nevertheless, one expects these two carbons to have a similar relaxation rate, being somewhat smaller than the one for the CC carbon because of the still larger distances involved. This is indeed observed. Finally, the relaxation rates should scale in proportion to R_1 of the α -CH₃ group; this agrees with the observed changes in i- and s-PMMA. We now also can explain why a plasticizer has so little effect on the observed relaxation rates. Even if the polymer backbone becomes much more mobile, the backbone motion still occurs at frequencies slow compared to the NMR frequency. Motion of the polymer backbone then contributes relatively little to the relaxation, as compared to the very effective relaxation contribution from the coupling with the α -CH3 protons.

The -OCH₃ carbons seem to be relaxed by their directly-bonded protons. The fast, free reorientation of the ester methyl protons does not yield a very effective relaxation mechanism, but the α -CH₃ protons are far away and contribute even less. This explains why R₁ is rather constant in all measured samples.

CONCLUSIONS

 13 C spin-lattice relaxation in solid PMMA yield under CP MAS NMR conditions, relaxation rates for each functional group. Except for the -OCH₃ relaxaton, the reorientation of the α -CH₃ group seems to be the dominant factor which determines the 13 C relaxation in the solid state. The relaxation proceeds slower in isotactic PMMA than it does in syndiotactic PMMA, which is a consequence of faster reorientation of the α -CH₃ group in the isotactic chain. The increase of mobility of the polymer backbone in the presence of a plasticizer is not reflected in a significant change of the spin-lattice relaxation rates R₁.

ACKNOWLEDGEMENTS

We greatly appreciate the help of Dr. J. Heijboer; of Dr. G. Challa, Ir. Y.Y. Tan, and Dr. F. Bosscher; and of Dr. J.G. Kloosterboer, in providing us the samples from sources 3,1, and 2, respectively.

REFERENCES

BAKAY, L., KURLAND, R.J., PARRISH, R.G., LEE, J.C., PENG, R.J. and BARTKOWSKI, H.M.: Exp. Brain Res. 23, 241 (1975) DODDRELL, D., GLUSHKO, V. and ALLERHAND, A.: J. Chem. Phys. <u>56</u>, 3683 (1972) GARROWAY, A.N., MONIZ, W.B. and RESING, H.A.: ACS Symp. Ser. <u>103</u> (Carbon-13 NMR Polym. Sci.) 67 (1979) HATADA, K., KITAYAMA, T., OKAMOTO, Y., OHTA, K., UMEMURA, Y. and YUKI, H .: Makrom. Chem. 178, 617 (1977) LYERLA, J.R., HORIKAWA, T.T. and JOHNSON, D.E.: J.Am.Chem.Soc. 99, 2463 (1977)LYERLA, J.R.: Contemp. Top. Polym. Sci. 3, 143 (1979) PEAT, I.R. and REYNOLDS, W.F.: Tetrah. Lett. 14, 1359 (1972) PINES, A., GIBBY, M.G. and WAUGH, J.S.: J. Chem. Phys. 59, 569 (1973) SCHAEFER, J., STEJSKAL, E.O. and BUCHDAHL, R.: Macromolecules 8, 291 (1975) SCHAEFER, J. and STEJSKAL, E.O.: J.Am.Chem.Soc. 98, 1031 (1976) SLICHTER, W.P.: J. Polym. Sci., Part C, 14, 33 (1966) STOLL, M.E., VEGA, A.J. and VAUGHAN, R.W.: Rev. Sci. Instrum. 48, 800 (1977) TORCHIA, D.A.: J. Magn. Resonance 30, 613 (1978) VanderHART, D.L. and GARROWAY, A.N.: J. Chem. Phys. 71, 2773 (1979) Van DIJK, P.A.S., SCHUT, W., Van OS, J.W.M., MENGER, E.M. and VEEMAN, W.S.: J. Phys. E: Sci. Instrum. 13, 1309 (1980) VEEMAN, W.S. and MENGER, E.M.: Bull. Magn. Resonance 2 (Proc. ISMAR-AMPERE Int. Conf. Magn. Resonance, Delft, 1980), 77 (1981) Van OS, J.W.M., and VEEMAN, W.S.: Rev. Sci. Instrum. 50, 445 (1979)

Received July 9 , 1981 Accepted July 20, 1981